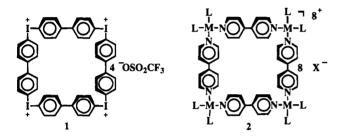
Hybrid, Iodonium-Transition Metal, Cationic Tetranuclear Macrocyclic Squares[†]

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There is considerable current interest in conformationally rigid, tetranuclear, macrocyclic ring systems with approximately 90° bond angles.¹⁻⁷ To date, two different types of macrocyclics have been reported: one type with covalent bonds and a main group element, iodine,⁶ at the corners, 1, and the other type with transition metals¹⁻⁵ at the corners held by chelating ligands as illustrated by 2. In this communication, we report the first example of hybrid main group iodonium-transition metal macrocyclic squares.



As more conventional recent methods^{8,9} of preparing the nitrogen-containing heteroaryl iodonium precursor 6 failed, we employed the procedures outlined in Scheme 1 for the formation of 6. Reaction of the bromophenylpyridine-derived anion 3 with vinyliodonium dichloride¹⁰ 4 gave bis(aryl)iodonium chloride 5 in 30-35% isolated yield. Conversion of 5 to the desired bis(aryl)iodonium triflate 6 proceeded smoothly in 95-98% isolated yield. Compound 6 is a stable, albeit somewhat hygroscopic, off-white microcrystalline solid.

Interaction of 6 with 1 equiv of the square planar chelated bis(triflates)³ 7 in acetone at room temperature gave the hybrid tetranuclear cationic macrocyclic squares 8, via self-assembly as shown in Scheme 2. Macrocyclic compounds 8 are remarkably stable, white, microcrystalline solids, soluble in organic solvents, and are fully characterized by analytical and spectral means as summarized in Table 1. In particular, the elemental analyses are consistent with the respective composition of 8. The ${}^{31}P{}^{1}H$ signals for 8 are sharp singlets shifted by 7.1 and 11.1 ppm, respectively, relative to the precursor³ metal triflates

[†] Dedicated to Professor Frank Albert Cotton on the occasion of his 65th birthday.

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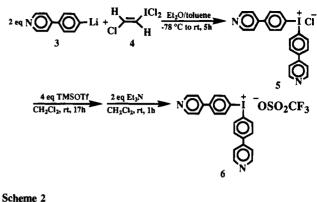
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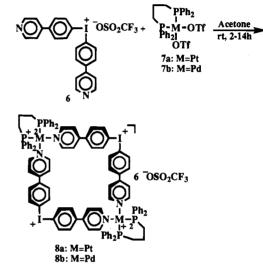
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(11) Anal. Calcd for 8a (C104H84F18I2N4P4Pt2O18S6'3H2O): C, 41.17;
H, 2.99; I, 8.37; N, 1.85; S, 6.34. Found: C, 41.38; H, 2.99; I, 8.33; N, 1.83; S, 6.32. Calcd for 8b (C104H84F18I2N4P4Pd2O18S6'6H2O): C, 42.91;
H. 3.32; I. 8.72; N, 1.93; S, 6.61. Found: C, 42.72; H, 3.05; I, 9.02; N. H, 3.32; I, 8.72; N, 1.93; S, 6.61. Found: C, 42.72; H, 3.05; I, 9.02; N, 1.99; S, 6.62.

Scheme 1





7. Especially diagnostic for the macrocyclic tetranuclear squares are the respective ¹H NMR signals. The phenylpyridine units show four multiplets, respectively, as expected for these symmetrical systems. Moreover, the signals α to the pyridyl nitrogen and α to the iodonium moiety are shifted downfield upon complexing, relative to the precursor iodonium salt 6. For **8a** and **8b**, the α -Py hydrogens are shifted by ~ 0.4 ppm and the α -iodonium units by ~0.05 ppm, relative to 6. In contrast the β -Py hydrogens are shifted upfield by approximately 0.2 ppm in both 8a and 8b relative to 6, presumably due to the shielding provided by the dppp phenyl rings of the chelating phosphine ligand on the metal. Integration of the proton signals is consistent with the relative intensities demanded by the macrocyclics, 8a and 8b. Also, as one would expect for such closely related congeners, the ¹H as well as ¹³C NMR spectra for 8a and 8b are very similar, whereas their ³¹P signals are significantly different. Furthermore, the spectral data of these novel hybrid macrocyclic squares closely track the respective NMR signals of the related pure iodonium bond macrocycle⁶ 1 as well as the pure transition metal bond macrocycles³ 2.

In summary, novel hybrid main group iodonium-transition metal, cationic, macrocyclic squares can be prepared in high yields via self-assembly from readily available precursors. These unique macrocycles add a new dimension to supramolecular¹² as well as potential host-guest¹³ chemistry.

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Table 1.	Physical and	Spectral I	Data for	Compounds 6	and 8

compd	yield (%)	mp (°C)	$IR (cm^{-1})^a$	¹ H NMR, δ (mult; <i>J</i> (Hz)) ^b	$^{13}C{^{1}H} NMR, \delta (mult; J (Hz))^{b}$	$^{31}P{^{1}H} NMR, \delta (mult; J (Hz))^{b.c}$	¹⁹ F NMR δ ^{b.d}
6	95–98	176-178 dec	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		C_{γ} -PhI), 137.43 (d, 11.2; C_{α} -PhI), 131.81 (d, 11.6;		-79.6 (s)
8a	73	253-256 dec	3089, 3052, 1617, 1437 (Ar), 1278, 1257, 1162, 1029 (OTf)	9.06 (bd, 5.0; 8H, H _{α} -Py), 8.42 (d, 8.7; 8H, H _{α} -PhI), 7.88-7.70 (m; 24H, H _{β} - PhI and H _{ϕ} -PhP), 7.53 (d, 6.7; 8H, H _{β} -Py), 7.48-7.34 (m; 24H, H _{m} - and H _{p} -PHP), 3.49 (bs; 8H, PCH ₂), 2.34 (m; 4H, PCH ₂ CH ₂)	151.76 (s; C_{α} -Py), 149.48 (s; C_{γ} -Py), 139.80 (s; C_{γ} -PhI), 137.88 (d, 4.1; C_{α} -PhI), 134.13 (bs; C_{σ} -PhP), 133.03 (s; C_{p} -PhP), 131.59 (d, 4.1; C_{β} -PhI), 130.19 (t, 5.6; C_{m} -PhP), 126.07 (m; C_{ipso} -PhP), 122.33 (g, 320.7; CF_{3}), 117.59 (s; C-I), 21.76 (m; PCH ₂), 18.52 (bs; PCH ₂ CH ₂)	-13.0 (s; $J_{Pt-P} = 3040$)	-78.0 (s)
8b	64	228-230 dec	3057, 1613, 1437 (Ar), 1277, 1256, 1156, 1028 (OTf)	9.03 (d, 6.4; 8H, H_{α} -Py), 8.41 (d, 8.8; 8H, H_{α} -PhI), 7.87-7.72 (m; 24H, H_{β} -PhI and H_{0} -PhP), 7.47 (d, 6.6; 8H, H_{β} -Py), 7.44-7.32 (m; 24H, $H_{m^{-}}$ and H_{p} -PhP), 3.38 (bs; 8H, PCH ₂) 2.37 (m; 4H, PCH ₂ CH ₂)	151.73 (s; C _a -Py), 148.93 (s; C _{\gamma} -Py), 140.18 (s; C _{\gamma} -PhI), 137.35 (s; C _a -PhI), 134.09 (t, 5.0; C _o -PhP), 133.00 (s; C _p -PhP), 131.55 (s; C _p -PhP), 126.63 (m; C _{lpso} -PhP), 124.77 (s; C _p -Py), 122.20 (q, 321.2; CF ₃), 117.34 (C-I), 21.78 (m; PCH ₂); 18.50 (bs; PCH ₂ CH ₂)	8.9 (s)	-78.0 (s)

^{*a*} Thin film, acetone. ^{*b*} NMR solvents: **6**, CD₃OD; **8a** and **8b**, acetone- d_6 . ^{*c*} Referenced to external 85% H₃PO₄ in a sealed capillary at 0.0 ppm. ^{*d*} Referenced to external CFCl₃ in a sealed capillary at 0.0 ppm.

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Supplementary Material Available: ¹H, ³¹P{¹H}, and ¹³C{¹H} NMR spectra for 6, 8a, and 8b (13 pages). This material is contained

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